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Sequential Metal-Catalyzed *N*-Heteroarylation and C–C Cross-Coupling Reactions: An Expedient Route to Tris(hetero)aryl Systems

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This paper describes copper-catalyzed N–C heteroarylation of benzimidazole, 1-methylbenzimidazolone, imidazole and pyrrole. The products of these reactions then undergo palladium-catalyzed C–C cross-couplings with aryl or heteroarylboronic acids under Suzuki–Miyaura conditions to provide a rapid entry, from readily-available reagents, into tris(hetero)-

aryl scaffolds comprising two or three N-heterocyclic rings. The sequential N–C and C–C couplings can be performed in a one-pot process (two examples are given: >50 % overall yields).

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Introduction

Arylated heterocycles are key structural motifs in a wide range of pharmaceuticals, agrochemicals, and organic functional materials, hence great effort has been devoted to their synthesis by a range of metal-catalyzed protocols.[1] Contemporary procedures can be readily performed with low catalyst loadings and good functional-group tolerance. These include C-heteroaryl bond formation by palladiumcatalyzed Suzuki-Miyaura cross-coupling reactions of arylboronic acids/esters with heteroaryl halides,[2] or by direct functionalization of heterocyclic C-H bonds with arvl halides^[1d] catalyzed by palladium^[3] or copper species.^[4] For heterocyclic N-arylation, which is relevant to the present work, the classical copper-promoted Ullmann reaction has been improved by the addition of a variety of ligands, such as diamines,^[5] pipecolinic acid,^[6] sterically hindered phosphanes,^[7] a mixture of 1,10-phenanthroline and dibenzylideneacetone,[8] 4,7-dimethoxy-1,10-phenanthroline,^[9] (S)-pyrrolidinylmethylimidazole, [10] N-hydroxyimides[11] or ninhydrin.[12] A CuCl-catalyzed N-arylation of imidazole with arylboronic acids has been developed in the absence of an additional chelating ligand. [13] Combined N(1)- and C(3)-arylations of 3-iodoindazole using boronic acids and Cu(OAc), have been reported. [14] However, these methods require additional steps to synthesize the arylboronic acid. You et al. have very recently shown that imidazoles can be N-arylated with aryl and heteroaryl halides in the presence of base and a catalytic amount of CuI: an excess of imidazole improved the product yield and it was noted that the imidazole substrate may also function as a ligand in this

process.^[15] The CuOAc-mediated *N*-arylation of indoles and carbazole with aryl iodides under base-free and ligandless conditions has been reported.^[16] New developments in palladium-^[17] and iron-catalysed^[18] *N*-arylations have also been reported recently.

While many important results have been achieved for *N*-arylations, it is notable that much less progress has been made for *N*-heteroarylations, i.e. analogous reactions with heteroaryl halides.^[15] The development of this methodology is, therefore, of considerable interest. Herein we present a series of heterocyclic *N*-heteroarylation reactions and exploit further reaction at an active halide site in the products via Suzuki–Miyaura cross-coupling reactions leading to the expedient construction of tris(hetero)aryl scaffolds comprising two or three N-heterocyclic rings.^[19]

Results and Discussion

We chose benzimidazole 1 and 1-methylbenzimidazolone 2 as starting reagents for the optimization of our strategy for two reasons: (i) their derivatives are widespread in structures of biological and medicinal importance; [20] (ii) their arylation reactions have been relatively neglected compared to other NH heterocycles; imidazole is most commonly used in this context. [3d,8,10,12,15,21] To confirm the versatility of the procedures, analogous reactions are also reported for imidazole (3) and pyrrole (4). The general concept of sequential N–C and C–C couplings leading to tris(hetero)aryl systems is depicted for benzimidazole in Scheme 1. The higher reactivity of iodides compared to bromides was exploited for selectivity in the reactions of the dihaloarenes in step 1. Scheme 2 depicts the N–C couplings and the specific examples are shown in Table 1.

In test reactions benzimidazole (1) reacted smoothly with 2-iodopyrimidine (5a) and 2-iodopyrazine (6) under modi-

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Scheme 1. Construction of tris(hetero)aryl scaffolds by sequential N-C and C-C couplings.

Scheme 2. Synthesis of 10-18. For conditions see Table 1.

fied Ullmann conditions (Cs₂CO₃, CuI, 1,10-phenanthroline, DMF, 110 °C)^[22] to afford the N-heteroarylated products 10 and 11 in 50% and 70% isolated yields, respectively (Table 1, Entries 1 and 2). The comparable reaction of 1 with 2-bromopyrimidine (5b) using the conditions developed by Cristau, Taillefer, et al.[5d] (Cs₂CO₃, Cu₂O, Schiff base ligand Chxn-py-al, acetonitrile, reflux) gave 10 in 78% yield. These workers found that the reaction of imidazole with 4-bromo-1-iodobenzene (1 equiv.) took place with complete regioselectivity at the iodine site. We have found that for reactions of the dihalo reagents 7 and 8 with the NH heterocycles 1–4 the optimum conditions to achieve selective displacement of the iodo substituent^[23] are: 1–4 (1.2 equiv.), 7 or 8 (1.0 equiv.) Cs₂CO₃ (2.0 equiv.) CuI (10 mol-%) and 1.10-phenanthroline (10 mol-%) in DMF at 80 °C (Table 1, Entries 3-6, 8 and 9). Higher reaction temperatures (e.g. 110 °C as used in Entries 1 and 2) resulted in considerably lower product yields for reactions of 7 and 8. The reaction of the less activated 4-bromo-1iodobenzene (9) with benzimidazolone 2 (Entry 7) required a higher temperature (100 °C) for efficient formation of 16. Table 1, Entries 3–9, show that moderate to high isolated yields were obtained for all the products 12–18, which possess a bromo substituent suitable for a subsequent C-C (hetero)arylation reaction. The reactions of 5-bromo-2iodopyridine (7) were especially efficient (Entries 3 and 5).

To accomplish step 2 of our strategy (Scheme 1) the compounds 12–18 were treated with the methoxyphenylboronic acid derivatives 19 and 20, and the pyridylboronic acid derivative 21^[24–26] in Pd-catalyzed Suzuki–Miyaura reactions^[2] to obtain the tris(hetero)aryl scaffolds 22–31 comprising two or three N-heterocyclic rings (Scheme 3). The alkoxy substituents on the aryl/heteroaryl boronic acids were chosen to ensure good solubility of the products. The results are collated in Table 2. Standard conditions using Pd(PPh₃)₂Cl₂ (ca. 5 mol-%) as catalyst, Na₂CO₃ (1 M) as base, in dioxane at 80 °C gave the products 22–26 in high

Scheme 3. Synthesis of 22–31. For conditions see Table 2.

Table 1. Catalytic N-arylation/heteroarylation (N–C bond formation)^[a].

Entry	HetNH	Ar–X	Product	Conditions, isolated yield (%) ^[b]
1		X—————————————————————————————————————	N N N N N N N N N N N N N N N N N N N	a, 50 (X = I) b, 78 (X = Br)
2	N N H	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	N N N N	a, 70
3	N N H	\bigs_Br 7	N N N Br	c, 90
4	N N H	I—N—Br N—8	N N N Br	c, 57
5	Me N N H		Me N O N N 14 Br	c, 94
6	Me N N O	N= N− Br 8	Me N N N N N Br	c, 76
7	Me N N H		Me N N N N N N N N N N N N N N N N N N N	d, 81
8	N N N N N N N N N N N N N N N N N N N	l—N= N−Br	N N N Br	c, 65
9	N 4	N—Br 8	N N N N N N N N N N N N N N N N N N N	c, 84

[a] Reaction conditions, a: **5a**, Cs_2CO_3 , CuI, 1,10-phenanthroline, DMF, 110 °C; b: **5b**, Cs_2CO_3 , Cu_2O , Schiff base ligand Chxn-py-al, acetonitrile, reflux; c: Cs_2CO_3 , CuI, 1,10-phenanthroline, DMF, 80 °C; d: Cs_2CO_3 , CuI, 1,10-phenanthroline, DMF, 80 \rightarrow 100 °C. [b] The quoted yields are for isolated product after purification by chromatography and/or recrystallization.

Table 2. Suzuki-Miyaura cross-couplings (C-C bond formation).[a]

Entry	HetN- Ar-Br	Ar–B(OH) ₂	Product	Conditions isolated yield (%) ^[b]
1	12	(HO) ₂ B OMe	N N N OME	a, 74
2	12	MeO (HO) ₂ B————————————————————————————————————	OMe 23	a, 96
3	12	(HO) ₂ B - N	OEt N	a, 96
4	13	EtO N (HO) ₂ B - 21	OEt N	a, 79
5	14	(HO) ₂ B 19 OMe	Me N O N N N N 26 OMe	a, 76
6	14	(HO) ₂ B N	Me N O OEt	a, 0 b, 67 c, 54
7	15	(HO) ₂ B N	Me N N OEt	a, 0 b, 43
8	16	EtO N (HO) ₂ B - 21	Me N O OEt	a, 30 b, 83
9	17	(HO) ₂ B - N	OEt N	a, 48
10	18	(HO) ₂ B — N	N OEt	a, 78

[a] Reaction conditions, a: Pd(PPh₃)₂Cl₂, 1,4-dioxane, Na₂CO₃ (1 M), 80 °C; b: Pd(PPh₃)₄, 1,4-dioxane, Na₂CO₃ (1 M), 80–90 °C; c: [Pd₂(dba)₃], 1,4-dioxane, K₃PO₄ (1.27 M), 100 °C. ^[b] The quoted yields are for the isolated product after purification by chromatography and/or recrystallization.

yields (Entries 1–5). To our surprise, applying these conditions to attempted reactions of **14** and **15** with boronic acid derivative **21** failed to yield any product – unreacted starting materials were recovered in near quantitative yield (Entries 6 and 7, conditions a). When the catalyst was changed to Pd(PPh₃)₄ (with Na₂CO₃ as base) or [Pd₂(dba)₃] (with K₃PO₄ as base)^[27] product **27** was obtained in 67 and 54% yields, respectively (Entry 6, conditions b, c). Similarly, conditions b gave **28** in 43% yield. A general trend was that the Suzuki–Miyaura reactions were less efficient in the benzimidazolone series, especially Entries 6–8, compared to benzimidazole, imidazole and pyrrole analogs (Entries 1–4, 9, 10).

Analogous one-pot, two-stage procedures gave compounds **25** and **28** starting from benzimidazole (1) and 1-methylbenzimidazolone (2) in 67% and 52% overall yields, respectively.

The molecular structure of **22** obtained by X-ray crystallographic analysis is shown in Figure 1.

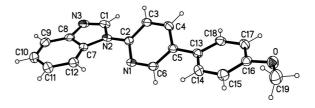


Figure 1. X-ray molecular structure of compound **22** (50% thermal ellipsoids). Dihedral angles [°]: benzimidazole/pyridine 30.7, pyridine/benzene 27.2, benzene/methoxy 10.2.

Conclusions

In summary, we have disclosed a series of reactions which constitute expedient and powerful methodology for the synthesis of new multi-heteroaryl scaffolds by a combination of N–C heteroarylation and C–C cross-couplings. Functionalized benzimidazole, benzimidazolone, imidazole and pyrrole derivatives have thereby been obtained. This protocol complements existing strategies which are of great importance in diverse areas of heterocyclic chemistry, especially the development of new molecular templates for drug discovery and materials chemistry applications. The practical benefits include readily-available and inexpensive starting materials, substrate versatility, experimentally straightforward procedures and good product yields. The sequential N–C and C–C couplings can be performed in a one-pot process in >50% overall yields.

Experimental Section

General: Details of equipment and techniques used are the same as those we have reported previously.^[28] All synthetic reagents were used as supplied. Solvents were dried and distilled using standard procedures. All reactions were performed under dry argon.

General Procedure for the *N*-Arylation/Heteroarylation Reactions of N-Heterocycles with Aryl/Heteroaryl Iodides: The halide (4.1 mmol)



and 1,10-phenanthroline (0.40 mmol) was added sequentially to a mixture of the N-heterocycle (4.8 mmol), CuI (0.20 mmol) and Cs₂CO₃ (8.0 mmol) in dry degassed DMF (4 mL). The mixture was stirred and heated under argon at 80 °C until TLC monitoring showed that the reaction was complete (typically ca. 24 h). The reaction mixture was cooled to room temperature, diluted with ethyl acetate (10 mL) and washed through a pad of silica gel with ethyl acetate (ca. 40 mL) to remove inorganic materials. The eluent was concentrated in vacuo and the product was purified either by recrystallization or by chromatography on a silica gel column.

1-(5-Bromopyridin-2-yl)-1*H*-benzimidazole (12): Chromatography (eluent EtOAc/Et₂O, 9:1 v/v) yielded 12 as a white solid in 90% yield, m.p. 159.6–160.5 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.61 (d, J = 2.0 Hz, 1 H), 8.51 (s, 1 H), 8.00-7.94 (m, 2 H), 7.86-7.84(m, 1 H), 7.44 (dd, J = 8.4, 0.4 Hz, 1 H), 7.39-7.32 (m, 2 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 150.7, 148.8, 145.0, 141.7, 141.3, 132.2, 124.7, 123.8, 121.1, 117.9, 115.4, 112.9 ppm. IR (\tilde{v}_{max} =): \tilde{v} = 3080, 3051, 1578, 1499, 1476, 1203, 741 cm⁻¹. MS (EI): m/z = 273 (100) [M⁺, ⁷⁹Br], 275 (97) [⁸¹Br]. C₁₂H₈BrN₃ (274.1): calcd. C 52.58, H 2.94, N 15.33; found C 52.53, H 2.83, N 15.19.

General Procedure for the Suzuki-Miyaura Cross-Coupling Reactions with Aryl/Heteroaryl Boronic Acids: Aqueous Na₂CO₃ (1 M, 4 mL) was added to a mixture of the aryl halide (1.8 mmol), the boronic acid (2.0 mmol) and catalyst (0.1 mmol) in 1,4-dioxane (8 mL). The mixture was stirred and heated at 80 °C until TLC monitoring showed that reaction was complete. The mixture was cooled to room temperature, solvent was removed in vacuo and EtOAc (50 mL) was added to the residue which was washed with brine (3 × 50 mL). The organic layer was separated, concentrated in vacuo and the product was purified by chromatography on a silica gel column.

1-[5-(2-Ethoxypyridin-3-yl)pyridin-2-yl]-1H-benzimidazole Chromatography (eluent EtOAc/dichloromethane, 10:1 v/v) yielded **24** as a white solid in 96% yield, m.p. 106.9–107.7 °C. ¹H NMR (400 MHz, CDCl₃): $\delta = 8.80$ (d, J = 2.0 Hz, 1 H), 8.62 (s, 1 H), 8.21 (dd, J = 4.8, 1.6 Hz, 1 H), 8.15 (dd, J = 8.8, 2.8 Hz, 1 H), 8.11 (dd, J = 7.2, 1.2 Hz, 1 H), 7.88 (dd, J = 7.2, 1.6 Hz, 1 H), 7.69 (dd, J = 7.2, 1.6 Hz, 1 HJ = 7.2, 2.0 Hz, 1 H), 7.63 (d, J = 8.0 Hz, 1 H), 7.42–7.34 (m, 2 H), 7.02 (dd, J = 7.2, 5.2 Hz, 1 H), 4.47 (q, J = 7.2 Hz, 2 H), 1.41 (t, J = 7.2 Hz, 3 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 160.9$, 149.7, 149.0, 147.3, 145.1, 141.7, 139.7, 138.6, 132.5, 131.3, 124.6, 123.6, 121.0, 120.3, 117.4, 113.8, 113.0, 62.5, 14.9 ppm. MS (EI) m/z 316 (M⁺ 100%). HRMS (ES+) calcd. for C₁₉H₁₆N₄O+H: m/z317.13969; found m/z 317.13975. IR: $\tilde{v} = 3141, 2975, 2901, 1589,$ 1494, 1457, 1230, 1037, 731 cm⁻¹.

1-[5-(2-Ethoxypyridin-3-yl)pyrimidin-2-yl]-1*H*-benzimidazole (25). One-Pot Procedure: 5-Bromo-2-iodopyrimidine (7, 0.570 g, 2.0 mmol) and 1,10-phenanthroline (0.072 g, 0.40 mmol) were added to a mixture of benzimidazole (1, 0.284 g, 2.4 mmol), Cs₂CO₃ (1.304 g, 4.0 mmol) and CuI (0.038 g, 0.20 mmol) in dry degassed DMF (4 mL). The mixture was stirred and heated under argon at 80 °C until TLC monitoring showed the reaction was complete (17 h). There was then added, under argon, (2-ethoxypyridin-3-yl)boronic acid (21, 0.367 g, 2.2 mmol), Pd(PPh₃)₂Cl₂ (0.084 g, 0.12 mmol), 1,4-dioxane (10 mL) and Na₂CO₃ (1 M, 4.4 mL) and stirring was continued at 80 °C for 17 h. The reaction was cooled to room temperature, the solvent removed in vacuo and EtOAc (100 mL) was added to the residue which was washed with brine (3×50 mL). The organic layer was separated, dried with MgSO₄ and concentrated in vacuo. The product was purified by column chromatography on silica (eluent DCM/EtOAc, 1:1 v/v) to yield 25 as a white solid (0.425 g, 67%) spectroscopically identical to the sample prepared by the two-step procedure, as described in the Supporting Information.

Supporting Information (see also the footnote on the first page of this article): Experimental methods (p. S2-S9), copies of NMR spectra (p. S10-S28.

CCDC-665603 contains supplementary crystallographic data for compound 22. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk./data_request/cif.

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